

THE METHOXYCARBONYLALKYLATION AND METHOXYCARBONYLALKYLIDENATION OF SILYL ENOL ETHERS<sup>1</sup>

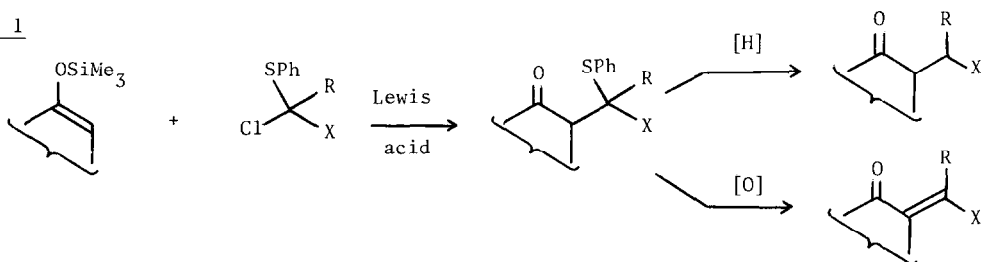
Ian Fleming\* and Javed Iqbal

University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW, England

*Summary*—Silyl enol ethers (**1**) react with the phenylthio(methoxycarbonyl)alkyl chlorides (**2**) in the presence of Lewis acids to give good yields of the  $\gamma$ -ketoesters (**3**); oxidative and reductive desulphurisations give the saturated (**4**) and unsaturated (**5** or **6**)  $\gamma$ -keto esters, respectively.

We reported earlier<sup>2</sup> that phenylthioalkylation of silyl enol ethers, using phenylthioalkyl chlorides and Lewis acid catalysis, was a powerful method for the regiocontrolled primary alkylation and alkylidenation of ketones and esters (Scheme 1). In our work so far the group X

Scheme 1

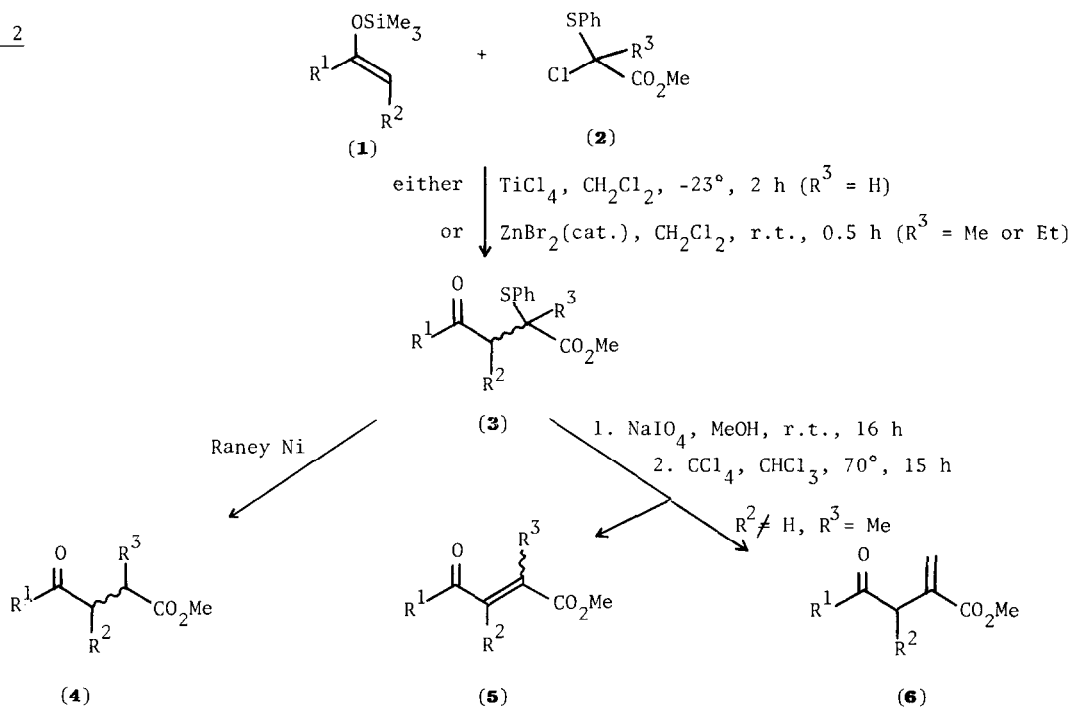


has been hydrogen.<sup>3</sup> We now report that the group X may be methoxycarbonyl, making the synthesis of saturated (**4**) and unsaturated (**5** or **6**)  $\gamma$ -ketoesters particularly easy (Scheme 2).<sup>4</sup> The reaction is regiocontrolled (entries 5 and 6 in the Table) and the starting materials (**2**) are easy to make by chlorination (NCS, CCl<sub>4</sub>, r.t., 10 h) of the corresponding  $\alpha$ -phenylthioesters. The reaction conditions are closely similar to those used by us earlier.<sup>2</sup> The oxidative desulphurisation (**3**  $\rightarrow$  **5** or **6**) usually gave the fully conjugated product (**5**), but when R<sup>2</sup> was not hydrogen and R<sup>3</sup> was a methyl group, the cycloelimination of phenylsulphenic acid took place towards the methyl group, and gave the  $\alpha$ -methylene ester (**6**). The alkylation step was not successful with silyl enol ethers derived from aldehydes or lactones.

NOTES and REFERENCES

1. No reprints available.
2. I. Paterson and I. Fleming, *Tetrahedron Lett.*, 995 and 2179 (1979); I. Fleming, J. Goldhill, and I. Paterson, *ibid.*, 3209; I. Fleming and T. V. Lee, *ibid.*, **22**, 705 (1981); I. Fleming and D. A. Perry, *Tetrahedron*, **37**, 4027 (1981); H. A. Khan and I. Paterson, *Tetrahedron Lett.*, **23**, 2399 (1982). See also, N. Ono, H. Miyake, and A. Kaji, *Synthesis*, 1003 (1981).
3. For secondary thioalkylation, see, M. T. Reetz and A. Giannis, *Synth. Commun.*, **11**, 315 (1981).
4. The combination of an  $\alpha$ -chloro- $\alpha$ -phenylthio ketone and a silyl enol ether has been shown to give furans: R. Tanikaga, K. Miyashita, H. Sugihara, and A. Kaji, *J. Chem. Soc. Chem. Commun.*, 1106 (1981).

Scheme 2



Table

Silyl enol ether (1)		Phenylthioalkyl chloride (2)	Yield %	Yield %	Yield %	Yield %
$\text{R}^1$	$\text{R}^2$	$\text{R}^3$	(3)	(4)	(5)	(6)
Ph	H	H	79	92	83	
Ph	Me	H	88	83	91	
Et	Me	H	78	81	90	
$(\text{CH}_2)_4$		H	88	91	85	
		H	69	93	81	
		H	78	83	-	
Ph	H	Me	83	77	78	
Ph	Me	Me	78	69		81
Et	Me	Me	90	83		81
$(\text{CH}_2)_4$		Me	90	87		81
EtO	Me	Me	77	73		76
Ph	H	Et	83		77	